

Potential for In Situ Treatment of Cyanide in Groundwater by Iron Addition

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Problem

- Iron cyanide complexes are obstinately uncooperative
- Very stable in groundwater and resists “destruction”
- Conventional pump and ex situ treatment is expensive

Typical Groundwater Remedy:

\$ Pump groundwater

\$ Ex situ treat in engineered treatment plant

\$ Dispose treatment residuals

Typical Groundwater Remedy:

\$ Very active management

\$ Well proven, reliable but expensive

\$ Ex. Cost \$25 million to treat 200 gpm
for 30 years

Potential In Situ Remedy:

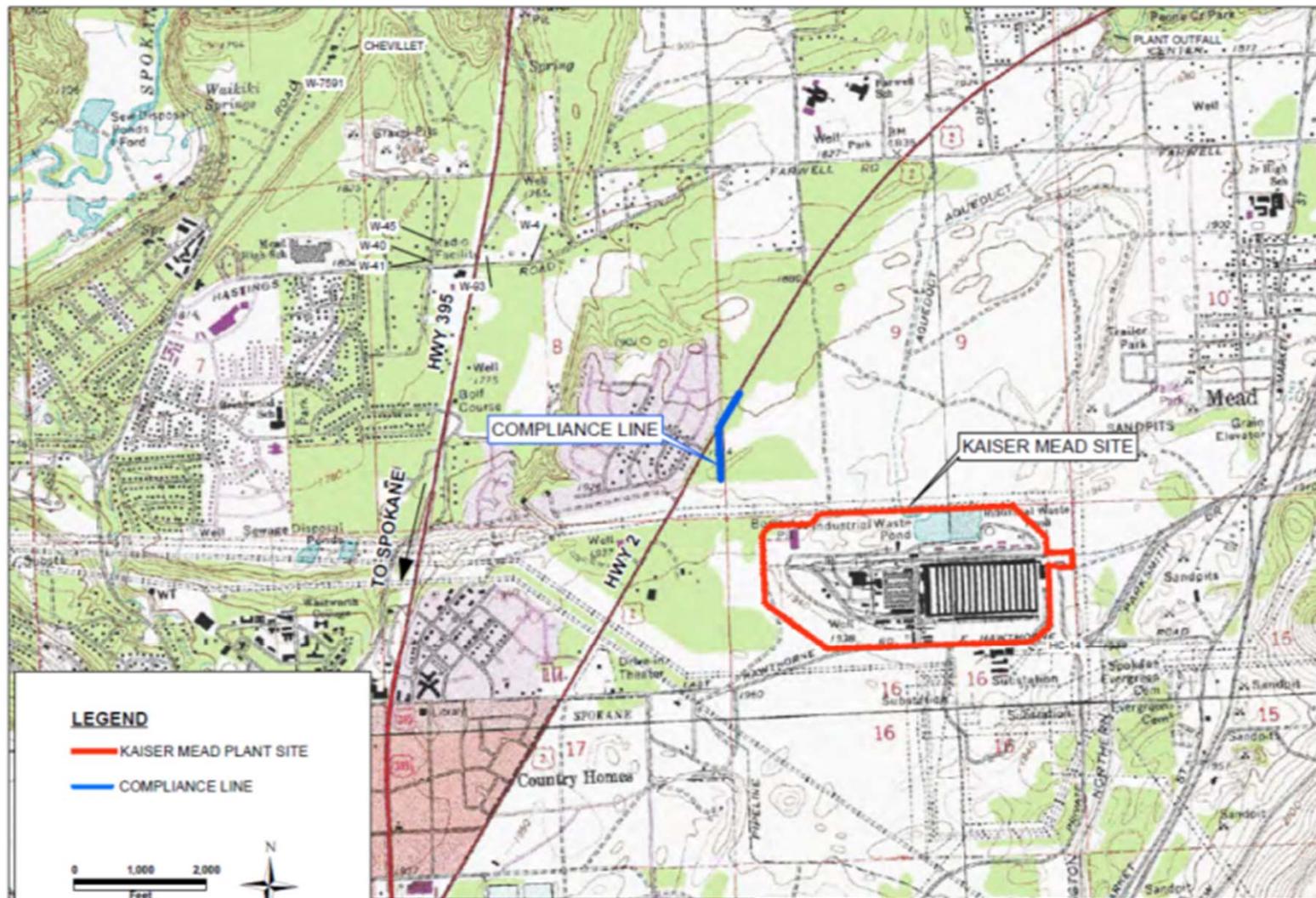
- Create &/or stabilize iron cyanide minerals within aquifer
- Reduce solubility of iron cyanide complexes
- Reduce Total and WAD CN concentrations in groundwater

Potential In Situ +/-

- + May be passive or low management
- + Cost fraction of ex situ treatment
- No full scale or long term field tests, so
- Long term reliability is unproven

Case Study: Kaiser Mead, WA Groundwater Cyanide Plume





KAISER MEAD NPL
MEAD WASHINGTON

SUPPLEMENTAL FEASIBILITY STUDY
SITE LOCATION MAP

FIGURE
3-1



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Natural Cyanide Sources

Numerous including:

- Biological - all plants, many microbes, even few arthropods.
- Forest Fires
- Lightning
- Carbonaceous meteorite impacts

Industrial Cyanide Sources

Numerous including:

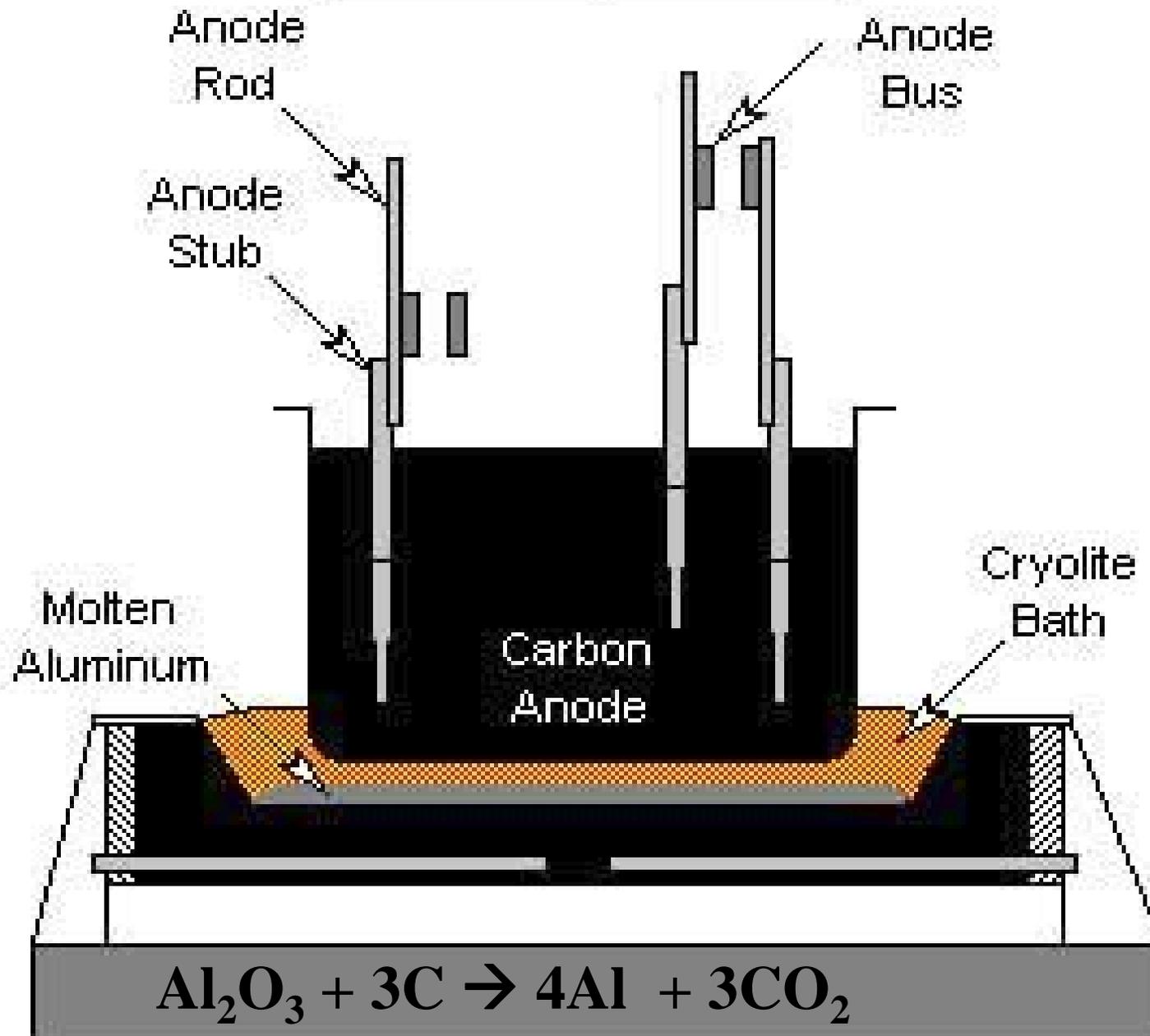
adhesives, electroplating, cement stabilizer, fire retardant, herbicides, pesticides, pharmaceuticals, & **hydrometallurgy**

Incidental Cyanide Sources

Numerous as produced during thermal processing of carbonaceous materials:

- Coking and coal gasification,
- Municipal waste sludge incineration
- **Alumina reduction**

Soderberg Method





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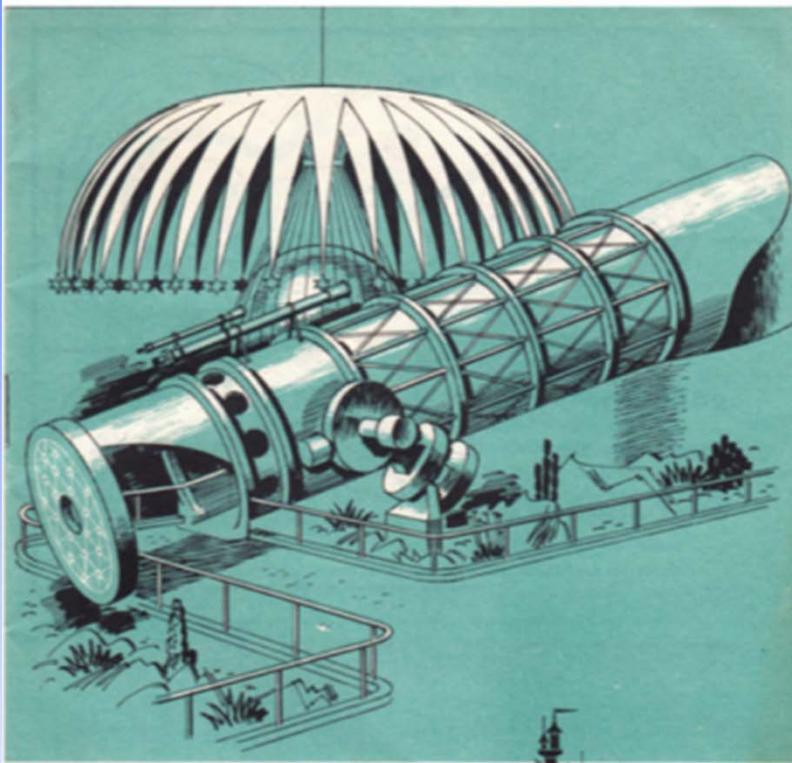
SPL Leachate Characteristics

Alkaline	pH 9 to 12
Fluoride	1,000 to 8,000 mg/L
Total Cyanide	200 to 4,000 mg/L
Ferricyanide	200 to 4,000 mg/L
WAD and Free CN	0 to 20 mg/L

Kaiser Mead Operational Timeline:

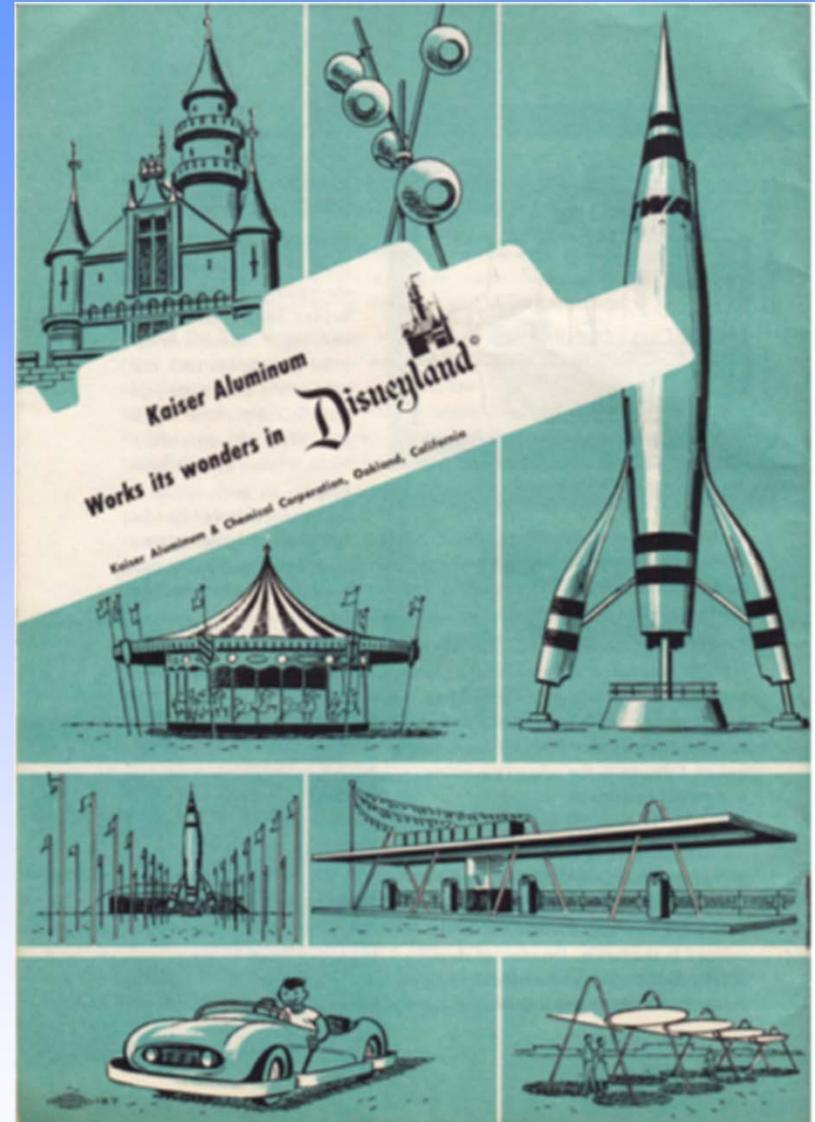
1942

Alumina reduction facility built by ALCOA for US government - High quality Al used to build airplanes



AT  Disneyland[®]
the Story of Aluminum

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Operational Timeline:

1946 ALCOA forced by US to divest due to fear of Al monopoly

1946 – 2000 Operated by Kaiser

2004 Kaiser bankruptcy & Consent Decree to resolve CERCLA liability

Asset Sales | Sale Details – 2-Day Onsite & Webcast Auction - Formerly 1.6MM SqFt Kaiser Aluminum Facility

- Details
- Preview/Removal
- Contacts
- Documents
- Terms & Conditions

Former Kaiser Aluminum Facility



Multi Million Dollar 2 Day Auction Sale. Over 3,000 Lots Equipment Across 182 Acres and in over 200 Buildings. Including a Complete Carbon Anode Plant, Machine and Metal Fabrication, Heavy Mobile Equipment, Plant Support Equipment, Millions of Dollars of Spares and MRO.

Sale Type	Sale Location	Currency	Online Bidding
On Site / Webcast Auction	2111 E Hawthorne Road Mead (Spokane), Washington, 99021 Map/Directions	USD	Click Here
Date & Time	Inspection / Preview		
Wed. May 2nd. & Thurs May 3rd. Start Time 9am (PST)	Tuesday May 1st (9am-4pm) Or by Appointment Contact: tracym@newmillcapital.com		
Buyer's Premium			
15% Onsite and 18% Online Buyer's Will Apply to All Sales in Addition to Applicable Local and State Sales Tax			

Featured Items

- [Photo Gallery](#)
- [Equipment List](#)
- [Brochure](#)
- [Lot Catalog](#)
- [Help](#)

- Related Sales**
- 

Onsite & Webcast Auction - Imperial Valley Cheese
 Wednesday | May 7, 2014
 El Centro, CA - Imperial Valley Cheese Swiss & Muenster Production & Packaging Facili... [Details...](#)
 - 

Real Estate Available - Imperial Valley Cheese
 El Centro, CA - Real Estate For Sale - 30,000 SqFt Dairy Products Manufacturing Plant... [Details...](#)
 - 

Webcast Auction - Assest No
 I onner Required by Gocner

Remediation Timeline:

1978 - CN and F found in drinking water wells

1983 - Placed on CERCLA NPL list

1983 – 2000 Interim remedial actions

2000 – Final Cleanup Action Plan Selected

Remediation Timeline

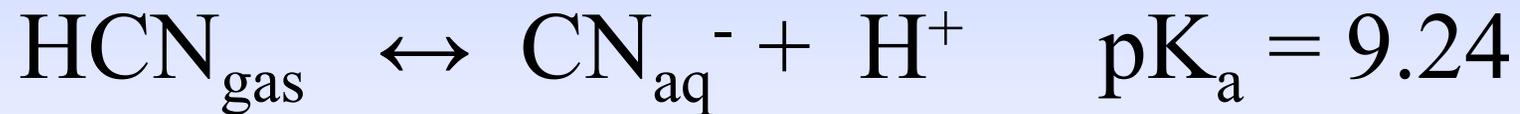
2002 Corrective Actions implemented –
(source control only)

2012 Performance Evaluation concludes
groundwater cleanup standards not met

2014 Draft Supplemental Feasibility Study

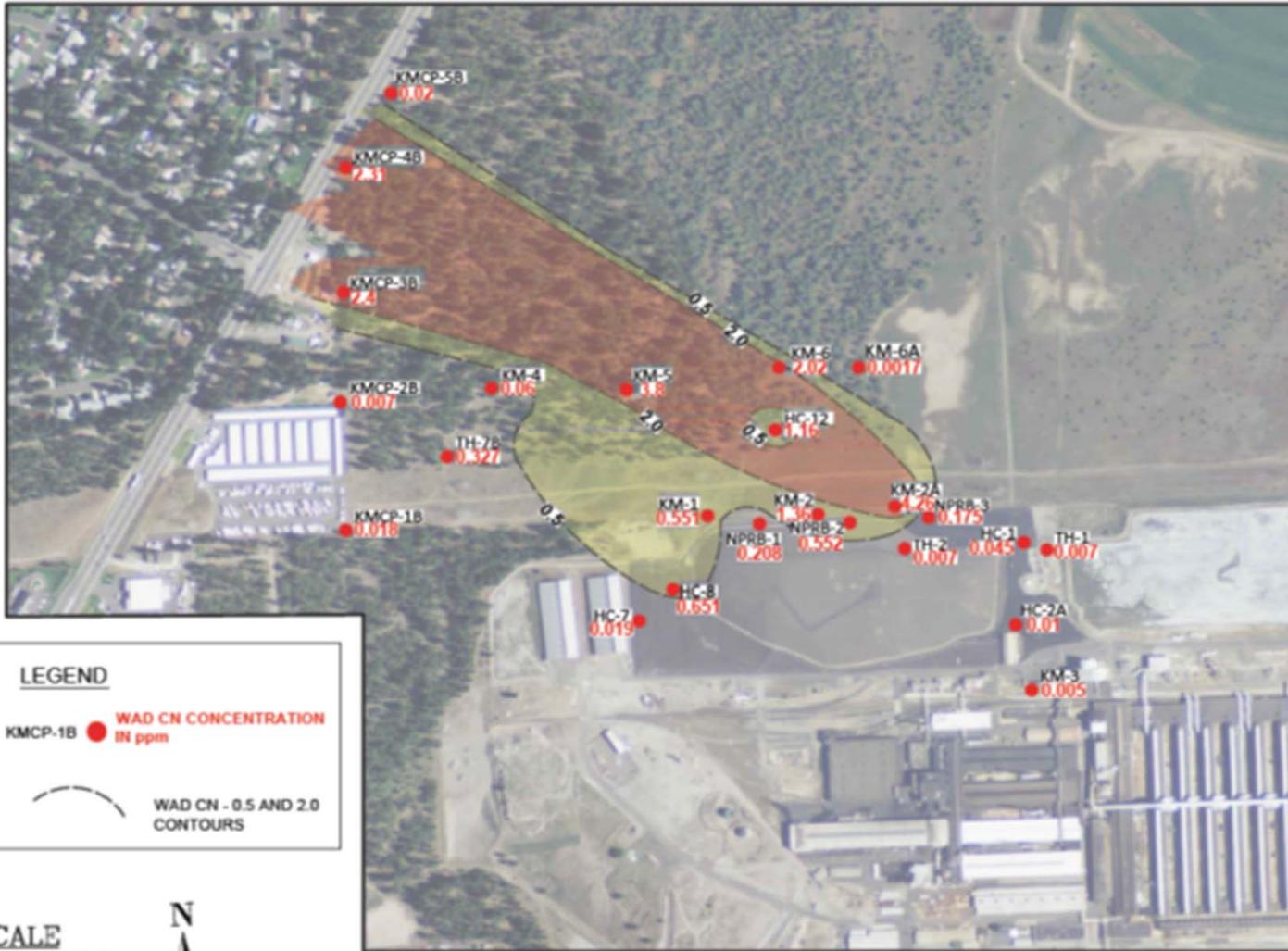
Aqueous Forms of Cyanide

“Free Cyanide”



WAD Cyanide Analytical Form

- “weak acid dissociable” = HCN gas released by weak acid digestion (pH 4.5 to 6)
- WAD = free CN + weak metal complexes (e.g. Ag, Cd, Cu, Zn)



SUPPLEMENTAL FEASIBILITY STUDY
KAISER MEAD NPL
MEAD, WASHINGTON

WAD CN CONCENTRATION
MAY 2013
KAISER MEAD

Total Cyanide Analytical Form

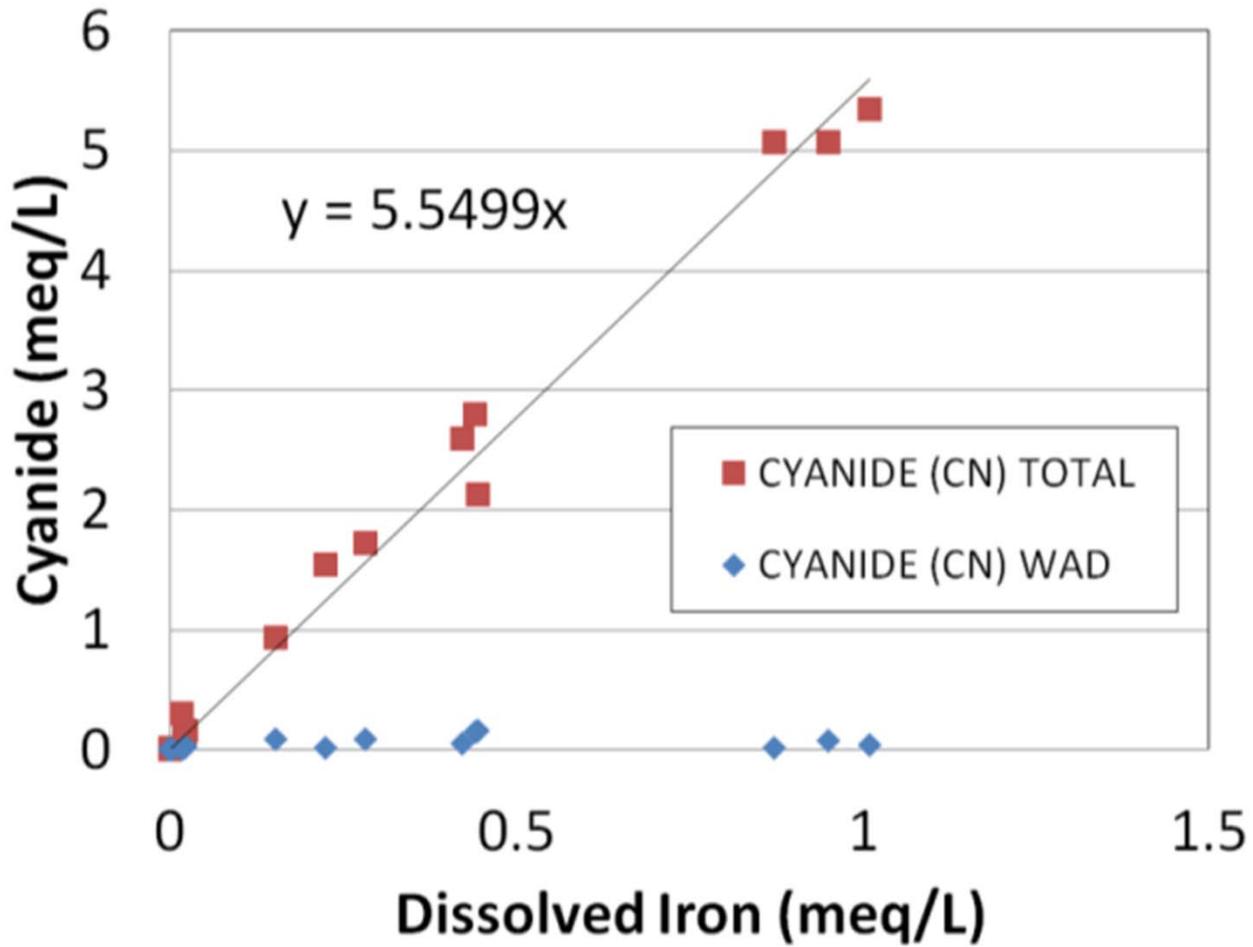
= HCN released by strong acid digestion
(pH < 2)

= WAD CN + strong CN complexes

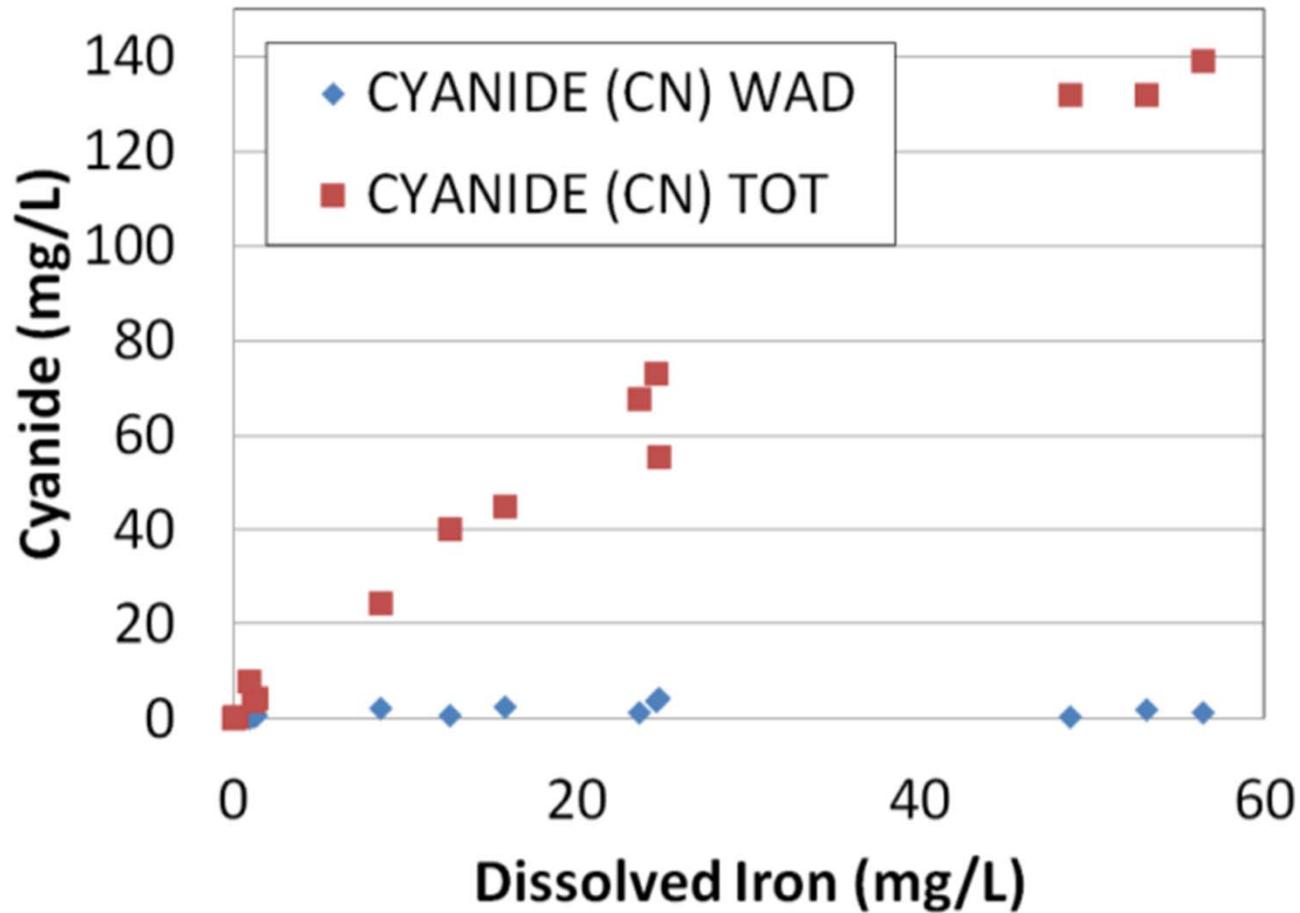
Strong Cyanide Complexes

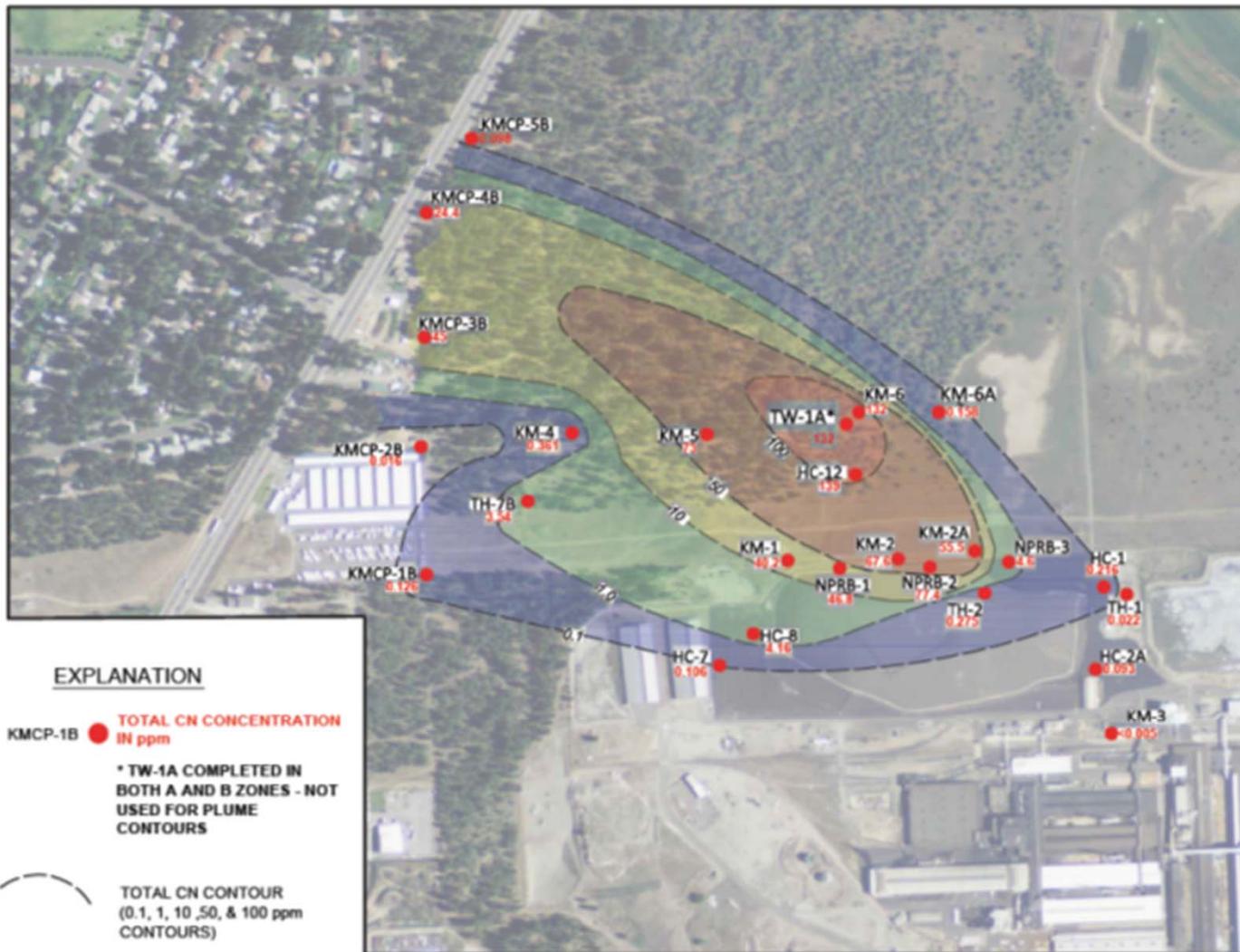
- Notably Iron and mixed Fe-metal-CN
(ex. $\text{Fe}(\text{CN})_6^{4-}$; $\text{KFe}(\text{CN})_6^{3-}$)
- But also Co, Pt, Au, Pd

Kaiser Mead Groundwater May 2013



Kaiser Mead Groundwater May 2013





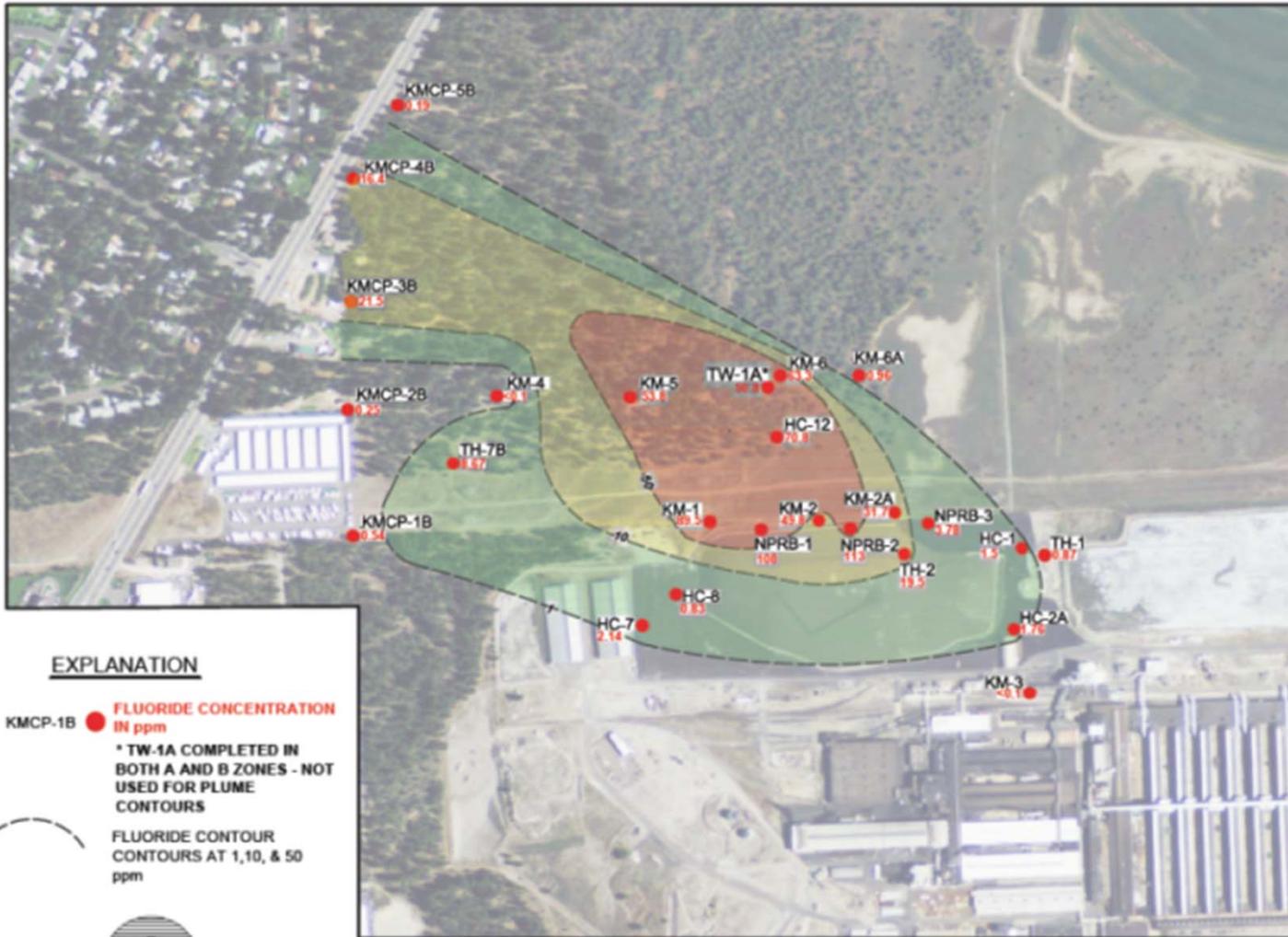
SCALE
(in Feet) 0 600



**TOTAL CN CONCENTRATION
MAY 2013
KAISER MEAD**



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EXPLANATION

KMCP-1B ● FLUORIDE CONCENTRATION IN ppm

* TW-1A COMPLETED IN BOTH A AND B ZONES - NOT USED FOR PLUME CONTOURS



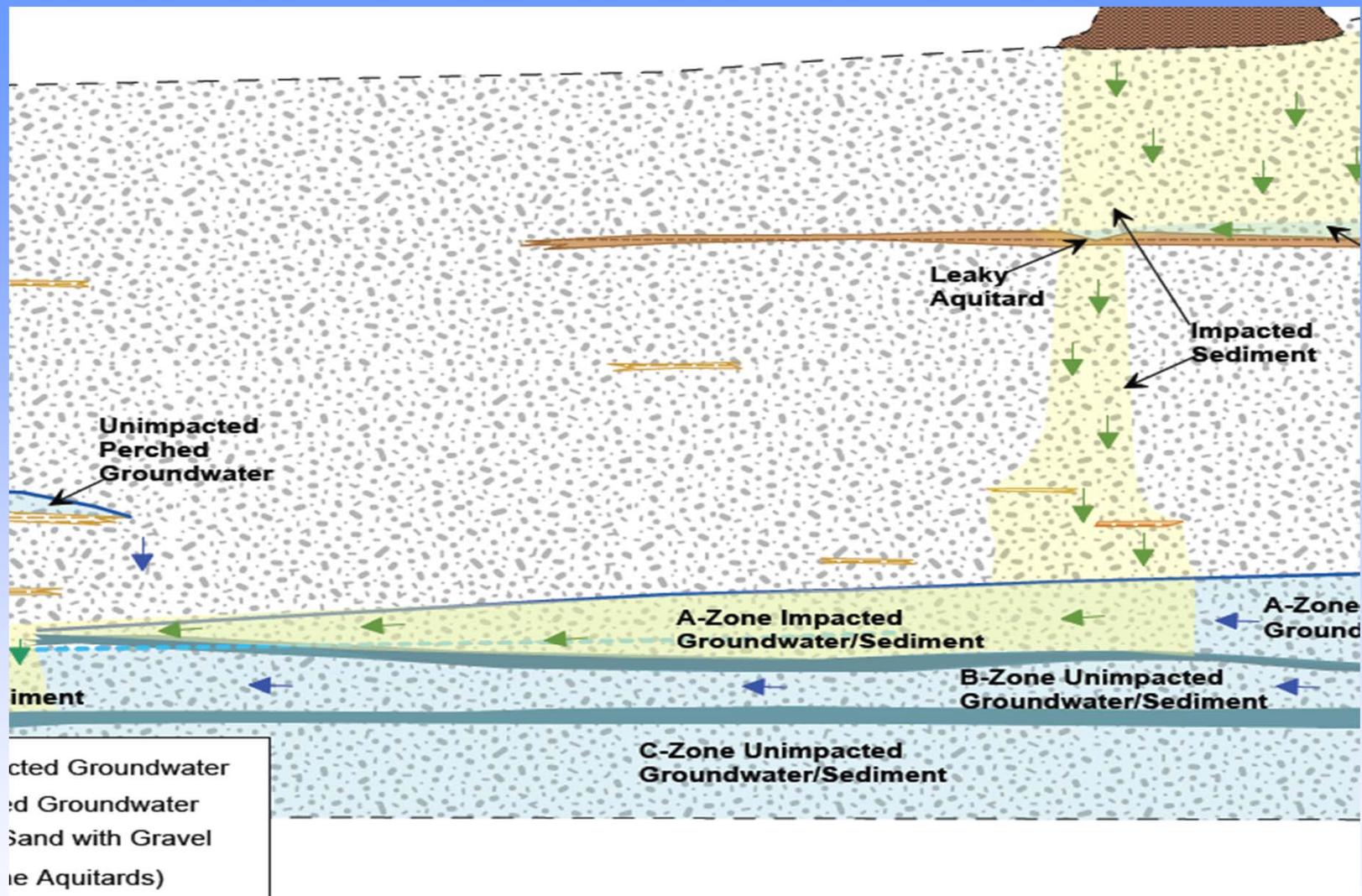
FLUORIDE CONTOUR CONTOURS AT 1, 10, & 50 ppm



SCALE
(In Feet) 0 600

**FLUORIDE CONCENTRATION
MAY 2013
KAISER MEAD**

FIGURE
2-3



cted Groundwater
 d Groundwater
 Sand with Gravel
 e Aquitards)

KAISER MEAD
CEPTUAL SITE MODEL

Conceptual Site Model

Interest In In Situ:

- Desire for lower cost option than P&T
- Evidence of secondary source of F & CN in vadose zone &/or aquifer suggest conditions may be right to further stabilize CN & F minerals within aquifer

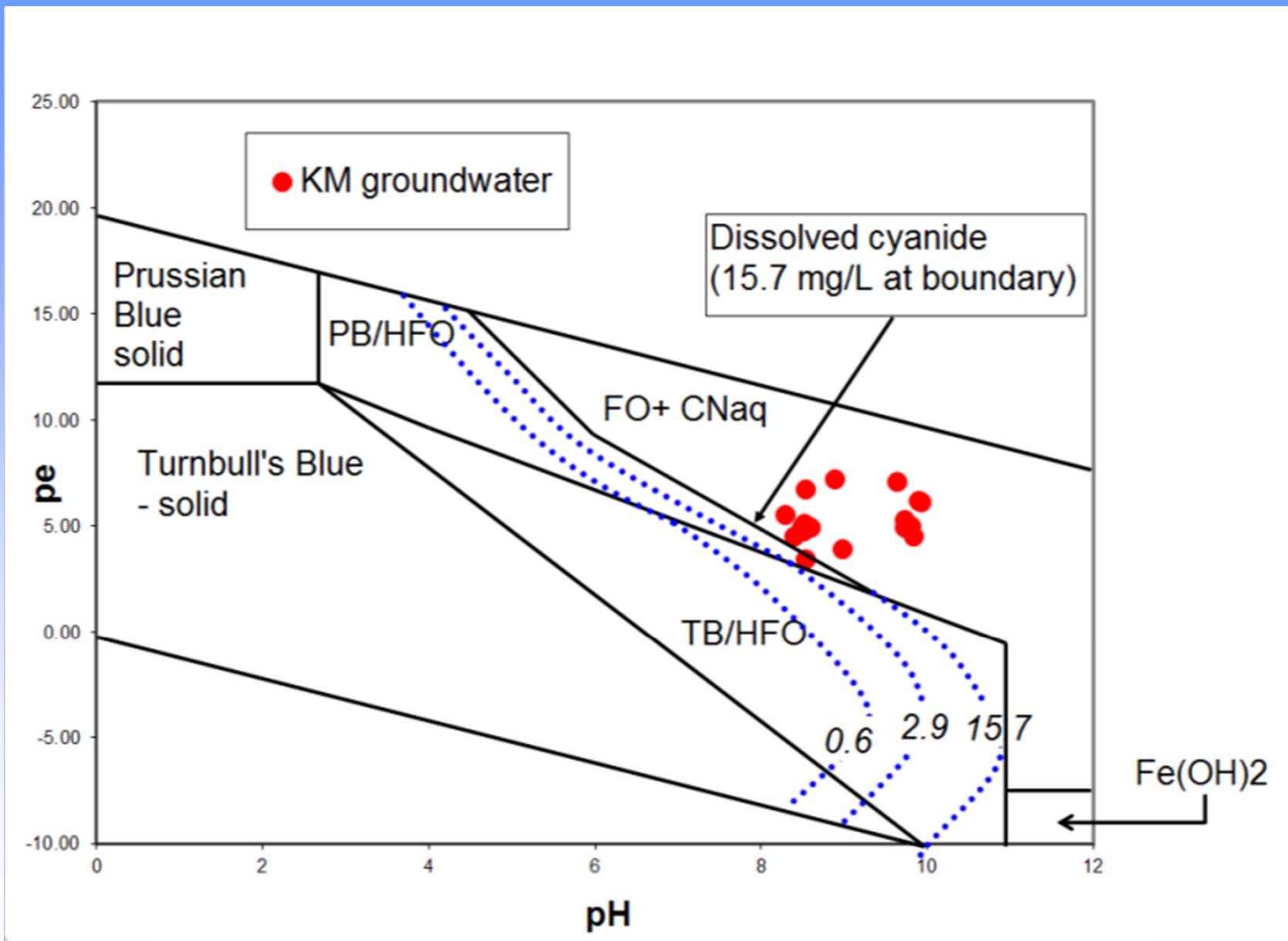
	CONDITIONS	RESULTS
Ghosh et al (1999)	Batch and column testing of PRB with ZVI	60 to 80 percent total cyanide removal
Wildeman et al (2006)	Testing of SRBRs with ZVI	ZVI alone did not remove CN but promoted biological removal
Peale et al (2008)	Bench and pilot-scale testing of PRB with carbon source and ZVI.	WAD CN reduced from 1 to 0.01 mg/L.

In Situ Concept

- Create &/or stabilize iron cyanide minerals within aquifer
- Reduce solubility of iron cyanide complexes
- Reduce Total, WAD and Free CN concentrations in groundwater

Iron Cyanide Minerals

- Prussian Blue : $\text{Fe}_4(\text{Fe}(\text{CN})_6)_3$
 - Acidic oxidizing conditions
- Turnbull's Blue: $\text{Fe}_3(\text{Fe}(\text{CN})_6)_2$
 - Neutral to alkaline reducing conditions
- Solid Solutions
 - Turnbull's Blue\Ferric Hydroxide ($\text{Fe}(\text{OH})_3$)
 - Prussian Blue\Ferric Hydroxide



Proof of Concept Testing



Two Reagents

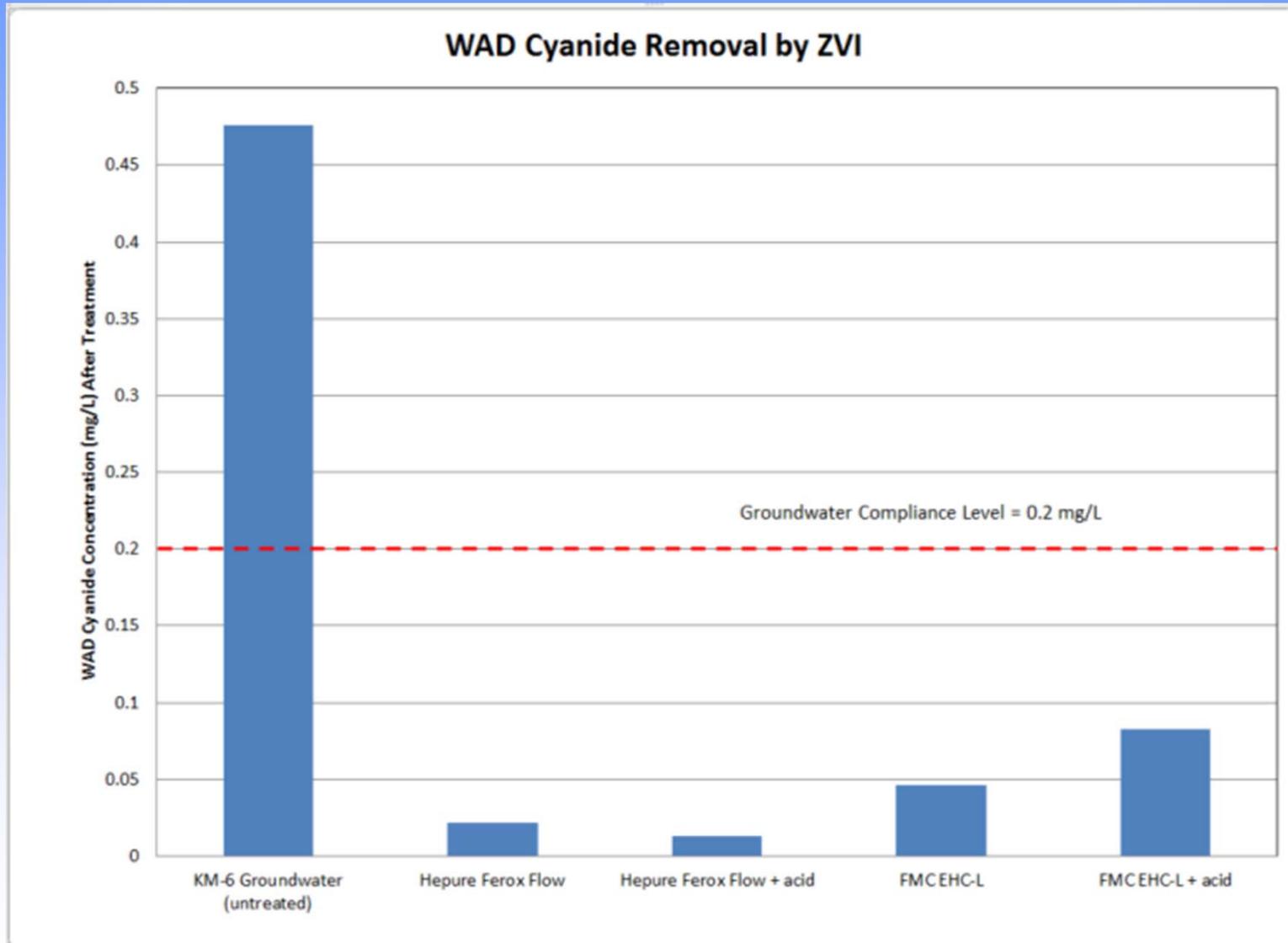
Hepure Technologies:

- FeroxTM Flow brand ZVI

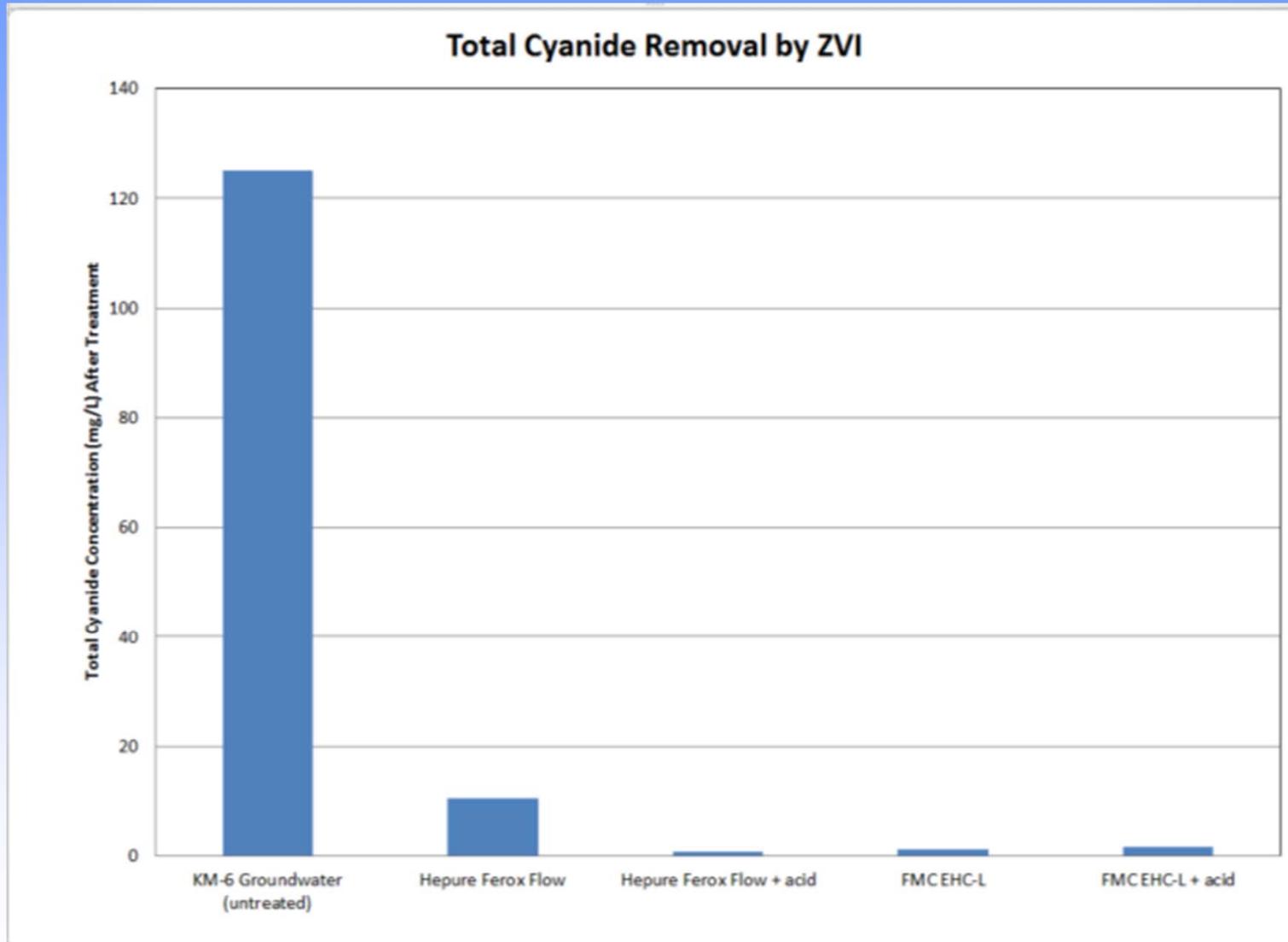
FMC Environmental Solutions:

- EHC-L emulsified ZVI with carbon and nutrients

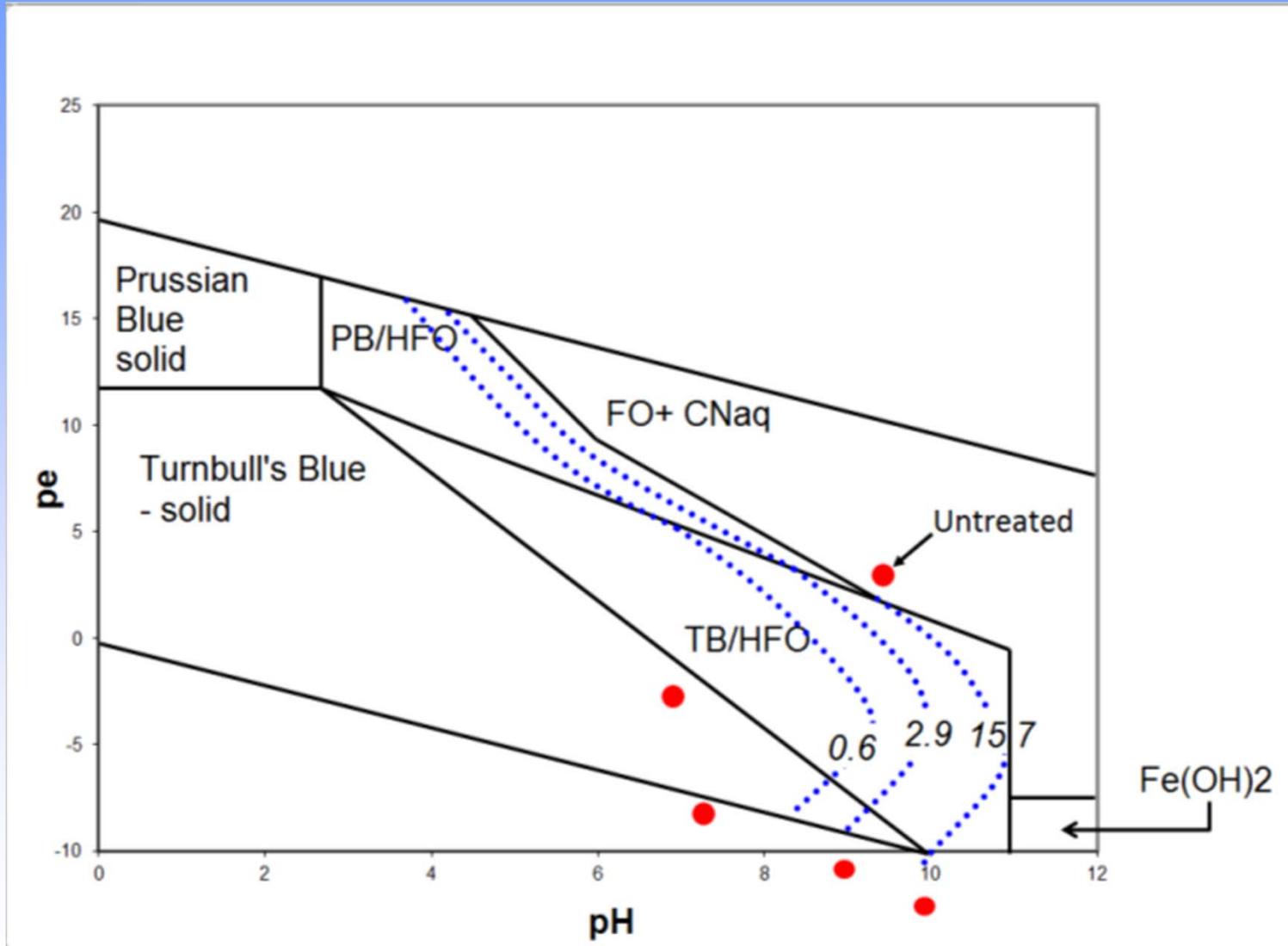
Results



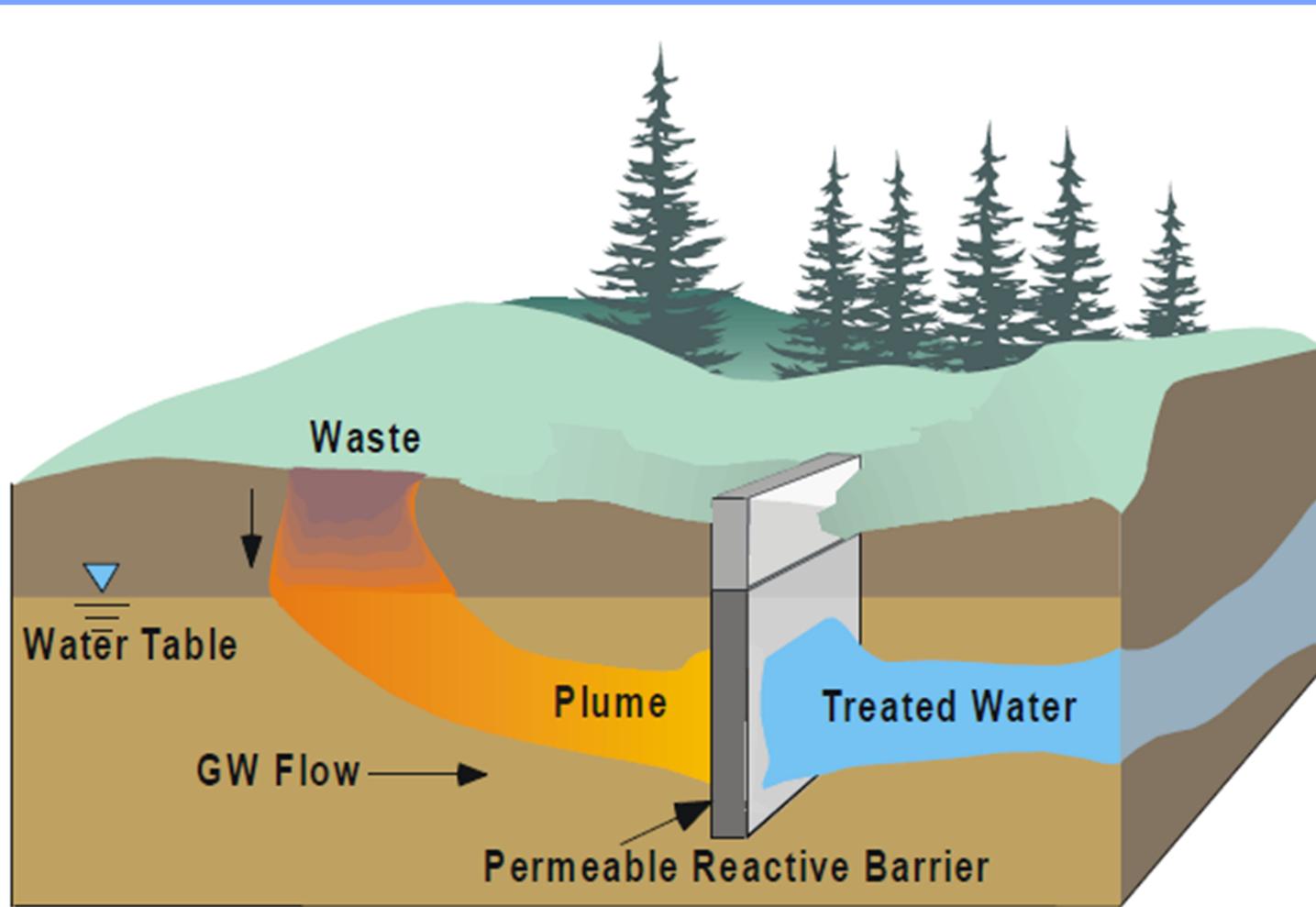
Results



Results



Implementation



Installation - Excavation



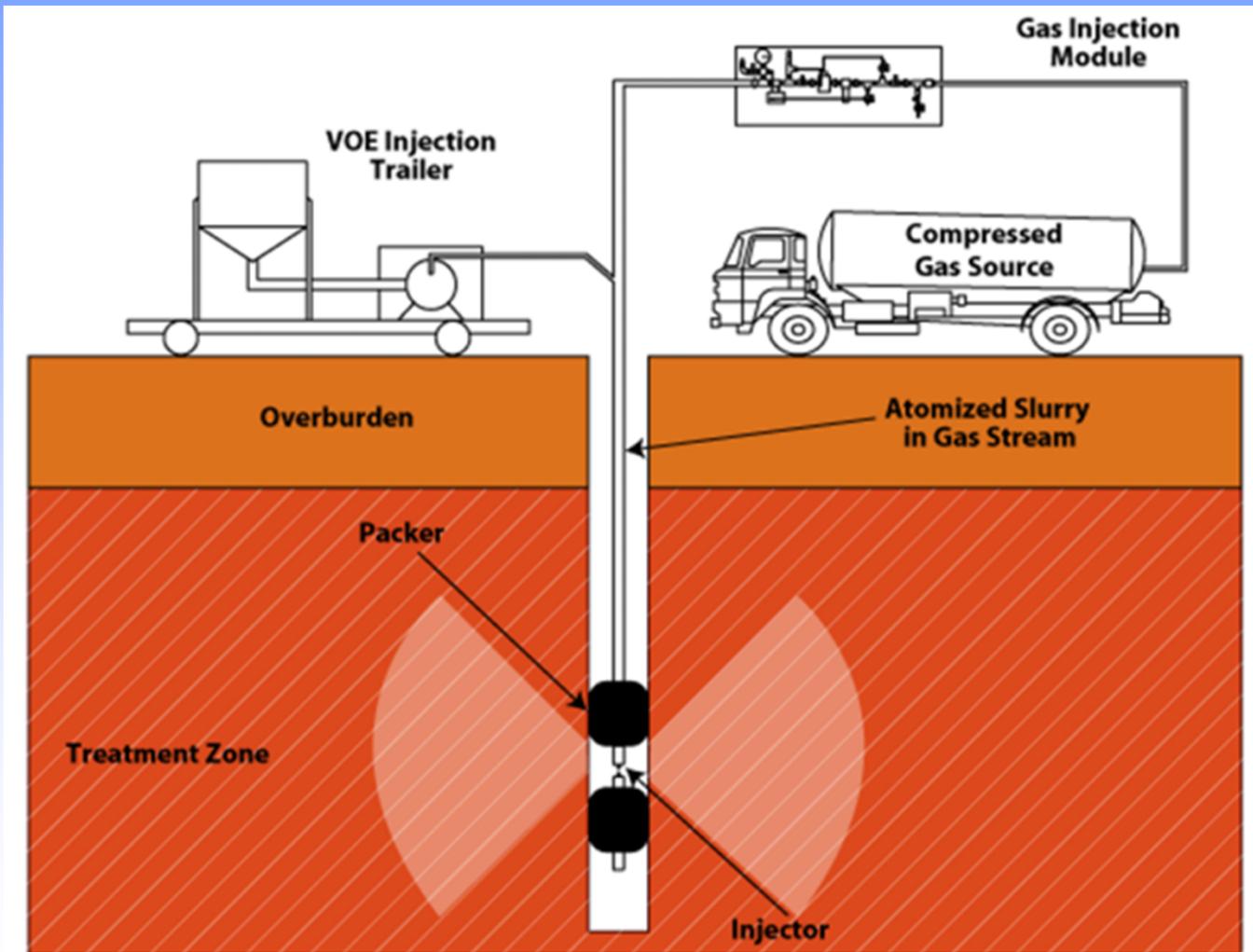
Installation - Trenching



Installation – Auger Mixing



Hydraulic Fracturing



Status of In Situ Treatment at Kaiser Mead:

- Concept is proven
- Rough cost approx. 20% of ex situ
- Uncertainties

Uncertainties

- No operating full scale systems
- Difficulty/cost of installation
- Hard to reverse/decommission once installed

Uncertainties

- Long term reliability and effectiveness:
 - Aquifer plugging
 - Reagent consumption
 - Noxious byproducts
 - Stability of cyanide minerals

Potential Best Fit for Sites:

- Total cyanide/iron cyanide complexes are particular concern
- Secondary CN source in subsurface
- Shallow aquifer

Potential Best Fit for Sites:

- Desire passive or semi-passive
- Desire minimal surface facilities
- Non-destruction of CN is OK

END



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